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solubility curve below Which salt is least soluble in water 2

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pogil common ion effect on solubility answers. According to the preceding exercise, the  $\text{Cu}^{2+}$  ion concentration in 1.0 M  $\text{NH}_3$  is only  $3.7 \times 10^{-14}\text{M}$ . In a system containing  $\text{NaCl}$  and  $\text{KCl}$ , the  $\text{Cl}^-$  ions are common ions.  $[\text{Cl}^-] = \frac{0.1\text{M} \times 10\text{mL} + 0.2\text{M} \times 5.0\text{mL}}{100}$ .

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