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370 Chapter 11 • Stoichiometry EXAMPLE Problem 11.1 Interpreting Chemical Equations The combustion of propane (C 3H 8) provides energy for heating homes, cooking food, and soldering metal parts. Interpret the equation for the combustion of propane in terms of representative particles, moles, and mass.

Chapter 11: Stoichiometry

Textbook pages: Chapter 12 Key Terms: stoichiometry, mole-mole problems, mass-mass problems, mass-volume problems, volume-volume problems. Directions: Use this information as a general reference tool to guide you through this unit. By the conclusion of this unit, you should know the following: Quantitative relationships exist in all chemical ...

CHAPTER 11: STOICHIOMETRY

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Chapter 11 Supplemental Problems The Mole Answer Key

We can see from the stoichiometry of the reaction that 3/2 mol of O 2 is required to produce 1 mol of H 2 SO 4. This is a standard stoichiometry problem of the type presented in Section 11.4, except this problem asks for the volume of one of the reactants (O 2) rather than its mass. We proceed exactly as in Section 11.4, using the strategy

Chapter 11.5: Stoichiometry Involving Gases - Chemistry ...

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